

# Design of Ion-Conducting Materials for High-Temperature Energy Applications

Author: Rajesh Keloth<sup>1</sup> Advisor: Shudipto Dishari<sup>1</sup>

<sup>1</sup> Department of Chemical and Biomolecular Engineering, University of Nebraska-Lincoln



## Introduction

- Acid-doped polybenzimidazole (PBI) membranes are promising for their reasonably high proton conductivities at high-temperature operations of proton exchange membrane fuel cells (PEMFCs) with little or no humidification<sup>1</sup>.
- However, PBI-based membranes have issues due to the weak acid-base interaction and phosphoric acid (PA) leaching<sup>2</sup>.
- Incorporation of quaternary ammonium group into the PBI matrix will improve PA retention due to strong ion pair interaction energy<sup>3</sup>.

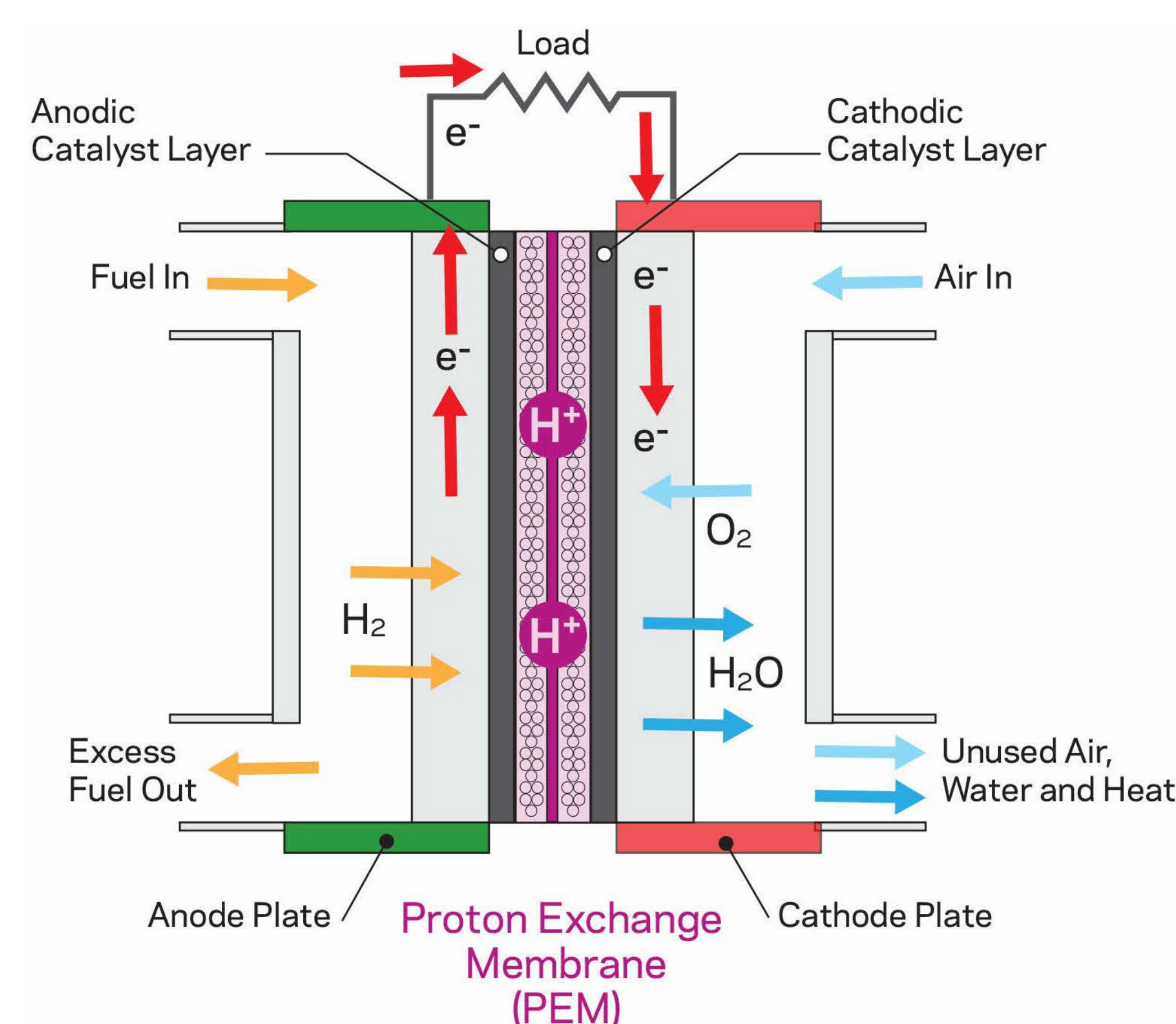
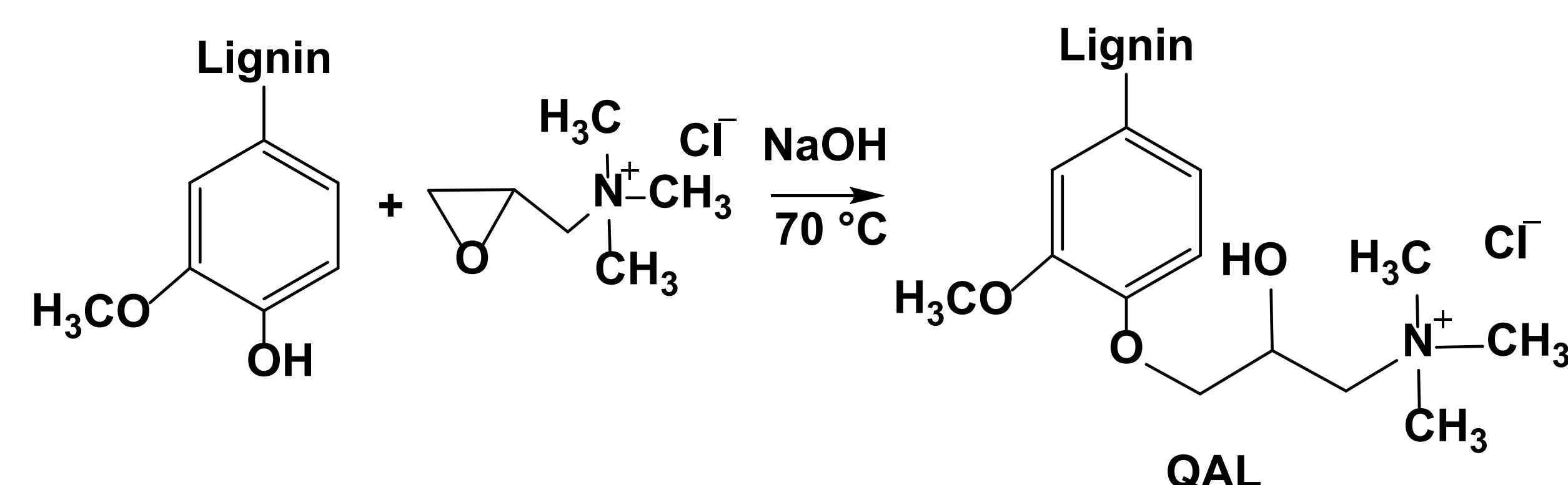


Figure 1. Working principle of PEMFC

## Research Objective

- Utilize lignin (a 3D-structured and branched polymer<sup>4</sup>) as a storage site for PA.
- Convert lignin's -OH groups to quaternary ammonium functionalities.
- These cationic functionalities will be utilized to hold PA within the PBI-QAL matrix via ion-pair interactions.

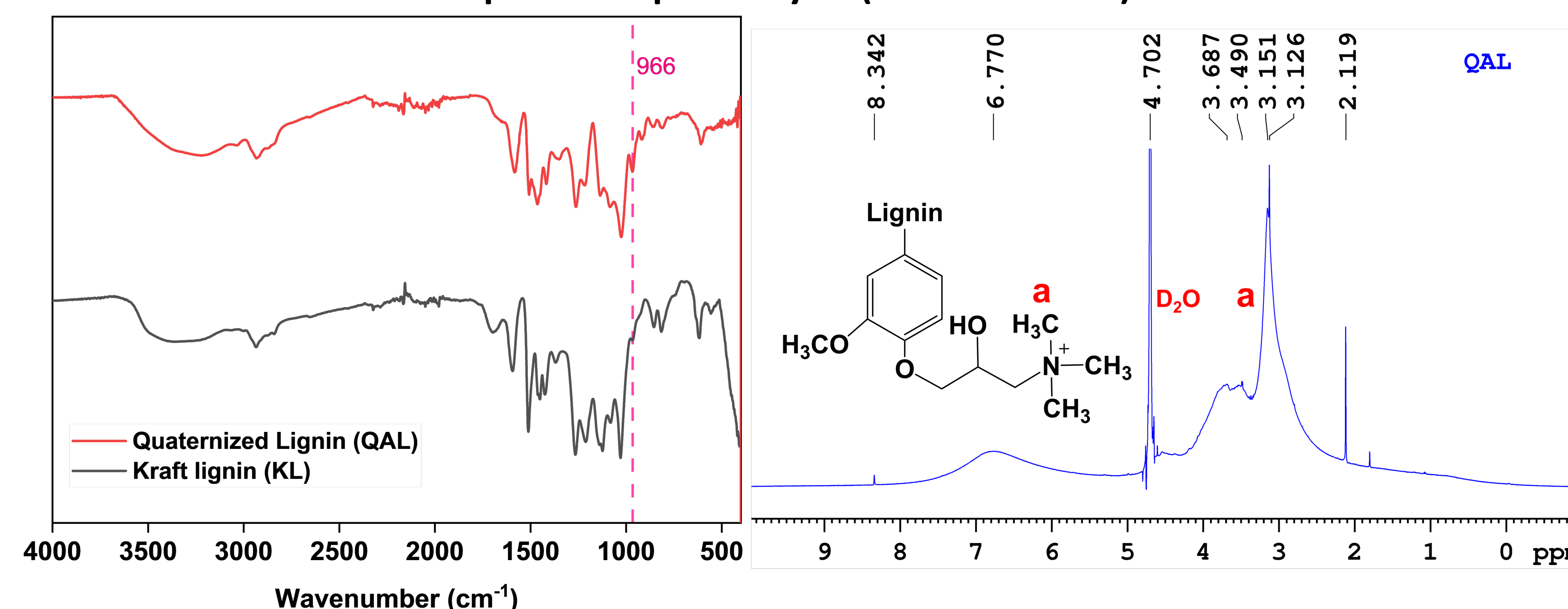
## Modification of Lignin



Scheme 1. Reaction scheme to render cationic functionalities to lignin and yield QAL<sup>5</sup>.

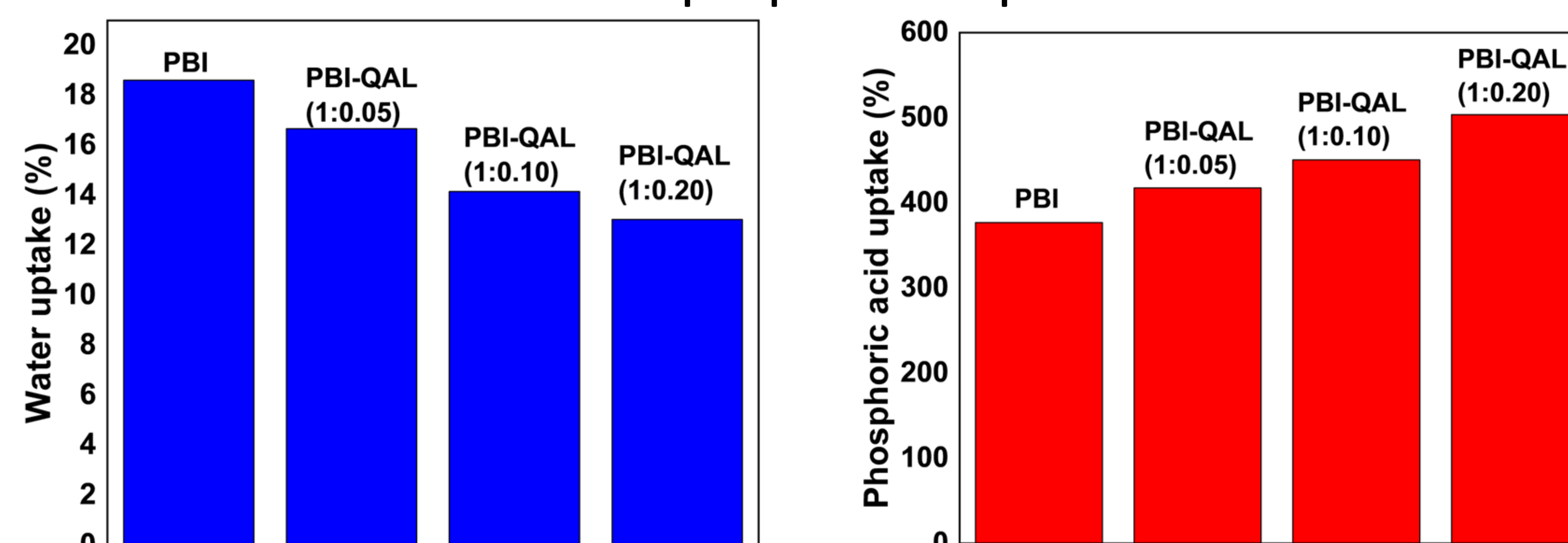
## Results

### Spectroscopic analysis (FTIR and NMR)



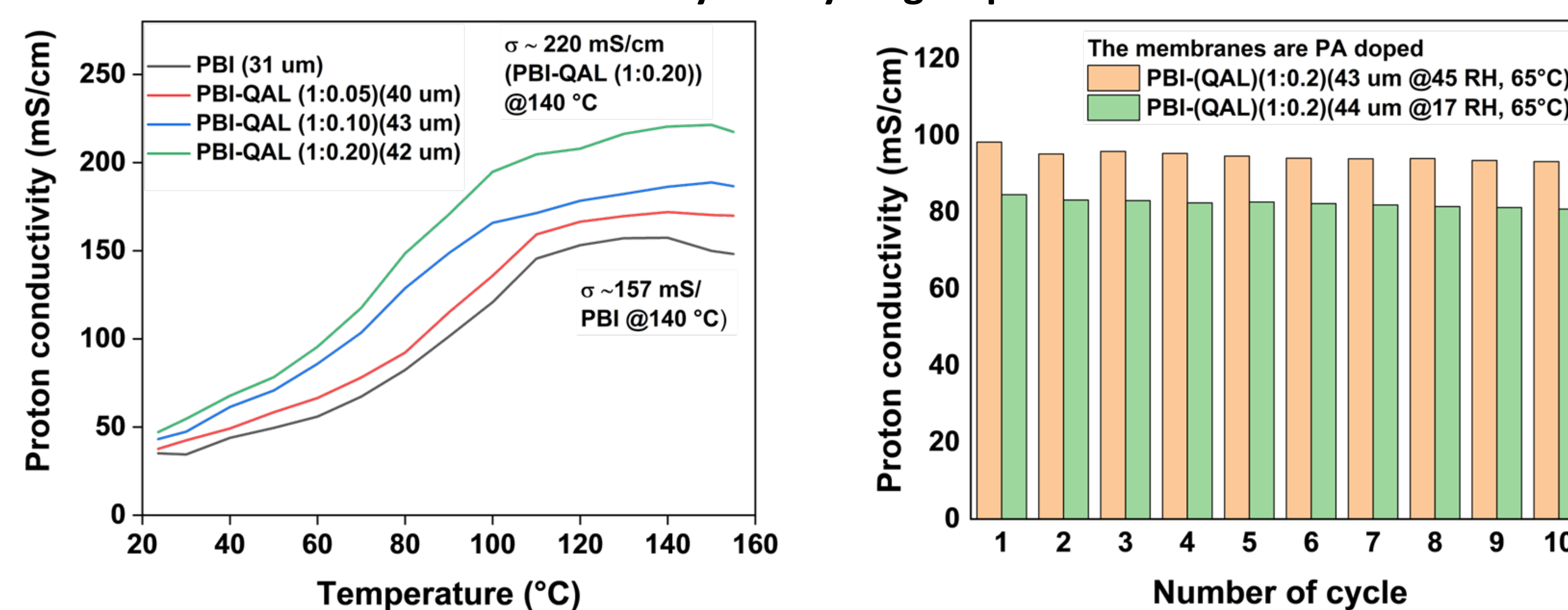
- The peak at 966 cm<sup>-1</sup> in QAL corresponds to CH<sub>2</sub> and methyl peaks of GTMAC.
- The peak range from 2.49 to 3.36 ppm corresponds to quaternary amine protons

### Water and phosphoric acid uptake



- Water uptake decreases as QAL content increases, due to the stiffness of QAL and its interaction with PBI.
- PA uptake of PBI-composite membranes was higher than that of pure PBI. The increase in acid uptake is a result of ion-pair interactions between PA and QAL.

### Proton conductivity and cycling experiment



- PBI membrane conductivity increased with increasing QAL loading.
- The cycling experiment showed a negligible reduction in conductivity over 10 cycles.
- This shows the strong ion-pair interaction of PA with PBI-QAL

## Conclusions and Future work

- Quaternary ammonium functionalities were successfully grafted on neutral kraft lignin.
- PA capture significantly improved with the increase in QAL loading in PBI membrane matrix.
- A significant improvement of proton conductivity at high temperature was achieved from PBI membranes upon QAL loading.

### Future Work

- Exploring cationic group incorporation techniques into lignin and study proton conductivity and cycling performance.

## References

1. Hooshyari, K.; Enhessari, M. J Power Sources 2015, 276, 62–72.
2. Matanovic, I.; Kim, Y. S. J. Phys. Chem. B 2020, 124, 7725–77.
3. Lee, A. S.; Kim, Y. S. J Mater Chem A Mater 2019, 7, 9867–987634.
4. Farzin, S.; Dishari, S. K. Front. Chem. 2020, 8, 1–17.
5. Kong, F.; Fatehi, P. Eur Polym J 2015, 67, 335–345.

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