

# Biosynthesis of Ethylene Glycol

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#### 1. Introduction

Fossil fuels are currently the primary source of energy around the world, with the biggest consumers being the transportation and industrial sectors. However, fossil fuels are finite and their rising prices along with their contribution to global warming has prompted research toward finding alternative forms of energy that are both sustainable and renewable. The development of biosynthetic routes to this of the test of the tes (EG). It is used as an antifreeze and coolant for automobiles and a desiccant for natural gas production, but its primary consumer is the polyester fiber production industry. Presently, the large portion of EG is produced via a synthetic route in which dimethyl oxalate obtained from nonrenewable fuels undergoes hydrogenation. In an effort to provide a means by which the increasing demands for EG can be met despite the dwindling supply of nonrenewable energy sources, we propose the use of a biosynthetic scheme that utilizes a carboxylic acid reductase (CAR) to reduce glycolate to glycoaldehde. Glycoaldehyde can be reduced to ethylene glycol through glycoaldehyde reductase (Fig 1.2). We are further motivated by research that has already shown the breadth of specificity displayed by Nocardia sp CAR and M. marimum CAR though the native substrate of CARs is benzoate (Fig 1.3). These CARs can reduce a wide range of substrates from aromatic to aliphatic compounds respectively. Our short-term goal is to identify a CAR variant with a high catalytic efficiency toward glycolate by testing CAR candidates from both reported and putative CARs of several organisms. Finally, through protein engineering we hope to evolve the best CAR candidate into a CAR with highly improved catalytic efficiency toward glycolate.



Fig 1.1 Seminal papers on CAR

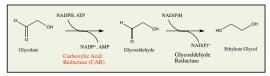


Fig 1.2 Proposed biosynthetic scheme for biosynthesis of EG

Applied and Environmental Microbiology, 2004; 70:1874-1881 Biotechnology Journal, 2014; 9, 822-843 Computational and Structural Biotechnology, 2014; 11:91-99 Chemical Reviews, 2013; 113, 4611-4632 PNAS, 2103; 110:87-92

### 2. CAR Requires Phosphopantetheinyl Transferase (sfp) for Catalytic Activity

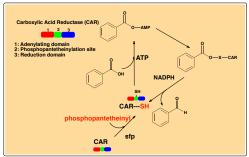
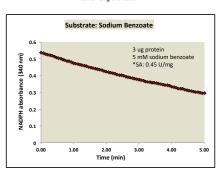


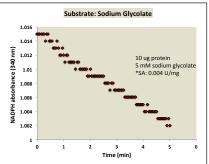
Fig 1.3 Nocardia sp. CAR (first reported CAR) scheme for catalysis of benzoate

### 5. Cloning and Expressing CAR Candidates

CARs were cloned into pET30b-sfp vectors with a His-tag at the C-terminal. The sfp encodes a phosphopantetheine transferase enzyme from Bacillus subtilis because previous has research has demonstrated that this post-transitional modification is necessary for CAR enzymes to become catalytically active (Fig. 1.3). Plasmids were transformed into BL21(DE3) and induced for protein expression in Luria-Bertani media with 0.5 mM IPTG. Cells were lysed by sonication and cell debris was removed via centrifugation. The supernatant loaded onto a column of Ni-sepharose resin, where the CAR-His-tag generally eluted with 5-10 mM imidazole

## 3. Nocardia sp. CAR Activity on Benzoate and Glycolate





The activity of Nocardia sp CAR on glycolate was low compared with its native substrate benzoate. Therefore, we sought to identify other CAR variants with improved catalytic efficiency.

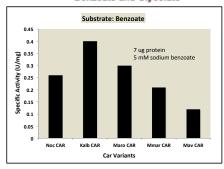
### 4. CAR Variants from NCBI Database

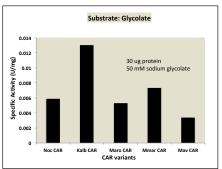
A BLAST search of the NCBI database was performed using the sequence of Nocardia sp. CAR. Hits were selected using a sequence identity >60% as the cut off for a total of 6 CARs (some reported and some putative):

- 1. \*Nocardia sp. 100% 2. \*M. marinum 62% 3. M. avium 61%
- 4. M. aromaticivorans 61%
- 5. R. wratislaviensis 62%
- 6. K. Albida 67%

\*Previously reported CARs

### 6. Activity of CAR Candidates on Benzoate and Glycolate





Preliminary data showing the activity of various CARs on benzoate glycolate

### 7. Conclusion

So far, we have tested the catalytic activity of CARs from 5 of the 6 candidates on glycolate. K. albida CAR is by far the most promising candidate and can catalyze the reduction of benzoate and glycoate. As far as we know, it is also the first CAR to have been reported from this strain.

### 8. Future Goals

We are colaborating with Dr. Mark Wilson to obtain the crystal structure of CAR so that we can use rational design to directly evolve the best CAR variant to have a high catalytic efficiency. We have also designed an unnatural amino acid to replace \$689 - the residue that is modified with posphopantetheinyl prosthetic group. This will allow the CAR to continually be in an "active" state and thus increase the rate of glycoate catalysis by CAR.

### 9. Acknowledgments

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