Conversion of CO₂ and Carbonates to Methane and (bio) Isoprene Merska CENTER FOR MERSKA CENTER MERSKA CENTER FOR MERSKA

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Abstract

We propose a biotechnology to simultaneously convert anthropogenic CO₂ or carbonate minerals from energy generation facilities into a biofuel (methane), and a bioproduct (isoprene), using microbial consortia. Carbon Capture and Utilization Strategies (CCUS) are critical to minimize emissions or remove anthropogenic CO, from the atmosphere. Yet, green technologies converting CO₂ to value-added products in addition to biofuels is lagging. To date, 30-40% of emitted CO, results from coal fired power plants and technologies have been developed to remove CO₂ from emissions. One of these technologies is the production of carbonate minerals such as calcium carbonate. Methane-producing microbial species in pure culture and in multi-organism microbial consortia are naturally capable of using anthropogenic carbonates or CO₂ for production of isoprene, which they incorporate into branched alkane lipids that constitute cell membranes. Isoprene is a valuable chemical commodity used in production of polyisoprene rubber (the major component of automotive tires), in styrene isoprene-sytrene (SIS) block copolymer adhesives, and as a synthetic intermediate for a wide range of specialty chemicals (cosmetics, vitamins, flavorings, etc). Microbially-derived isoprene is chemically identical to petroleum-derived isoprene, but can be extracted in higher yield and purity. This technology will be developed in two objectives: (Objective 1) quantify conversion of CO₂ and carbonates to methane and isoprene and (Objective 2) engineer methane-producing microbes to overproduce isoprene. These experimental objectives will be initiated in Year One in an effort to augment natural isoprene production in new and existing methane-producing microbial isolates or consortia. This technology will result in an inexpensive, economically feasible method to convert anthropogenic CO₂ and carbonates into an energy source and a commodity with increasing worldwide demand. Successful implementation will also result in a market for carbonates captured by power plants.

Introduction

Methanogens
Obligate anaerobes found in digestive tracts of animals, rice paddies, and wetlands.
Produce all bio-methane worldwide
Isopreniod-based membrane produced via Mevalonate Pathway
Methanogen lipids used in industrial lubricants and vaccine delivery systems
Organisms:
Methanosarcina acetivorans (M. ace) Methanosarcina barkeri (M. bar)

Isoprene • 2-methyl-1,3-butadiene • Produced mainly by plants • Component of synthetic Rubber



Figure 1: Methanogen Habitats: Cattle (A), Rice Paddies (B), and Freshwater Methanosarcina (C)



Table 1: Uses of Isoprene



Figure 4: Theresa Street Wastewater Treatment Plant Lincoln, NE. Pictured are large anaerobic digesters which have the capacity to facilitate methanogen growth and biomethane production.

Figure 3: Project Overview

 $CO_2 \rightarrow C$

bioisoprene

Figure 3: Gerald Gentleman Station. Located in Southerland, Nebraska, this coal fire electrical plant is the largest generating plant in the state. Operating at full capacity this facility could produce up to 20 million tons of CO_2 a year. The carbonate from this output is a potential source of carbon for methanogen



1,300,000 \$4.3 B

Gene A



- Figure 7: Plasmid Map for pJA1 with Gene A Insert
- Cloned and transformed plasmid into *E.coli* for production of transfectable plasmid
- Transfected plasmid JA1 into
 Methanosarcina acetivorans and
 Methanosarcina barkeri using liposomal
 NB118

 Verified insertion of gene of interest into genomes by PCR

Methane Production from 100 mL culture





Growth Curves of Gene B vs WT

60 80 Time (hours) ----M. bar WT -----M. bar Mutant

100 120

Figure 9: End-Point Methane Production assay 100 mL cell cultures Cells grown to early stationary phase

Gas Chromatography analysis of headspace

growth.

No significant difference in amount of methane produced between *M. acetivorans* WT and *M. acetivorans* Mutant, however significant overproduction of methane when comparing *M. barkeri* WT to *M. barkeri* Mutant.

Figure 10: Gene A Activity Assay • Cells grown to mid-/late- exponential phase • Whole cell lysates

Carbonates









Figure 8: Methanogen Membrane Lipid Structures: soprenoid ethers, backbone of comples lipids of methanogenic archaea. (a) Di-biphytanyl-diglycerol-tetraether; (b) macrocyclic diether; (c) tetritol-diether; (d) 3-O-phytanyl-sn-glycerol

EnzChek Pyrophosphate Assay Kit Increase absorbance at 360 nm

Successful expression of gene A in *M. acetivorans*

Figure 11: Growth Curve of Methanogen Transfectants • Cells grown in quintuplicate • OD₆₀₀ taken every 4 hrs

M. barkeri Mutant exits lag phase quicker when growing Exhibits different growth profile

Figure 16: Calcium Carbonate Mineral in the form of Calcite.

Implications

нс=с

Figure 17: Deep Core Sonic Drill.



Figure 17: Various renewable industrial products which can be produced by engineered methanogens.

нс=



Gene B



Figure 12: Plasmid Map for pJA2 with Gene B Insert

Gene of interest codon optimized for methanogen transcription

Cloned and transformed plasmid into E. coli for production of transfectable plasmid

• Transfected plasmid pJA2 into *M. acetivorans* and *M. barkeri* using liposomal transfection

• Expression of Gene B under control on high constitutive promoter proved to be lethal to both species when plated on HS agar plates



Future Directions

Gene A

Methane Rate Assay
Quantitative Lipid Profile analysis using HPLC
Full characterization of enzymatic activity

Gene B

Obtain stable transformants

Isoprene detection using Gas Chromatography

Characterization of promoter systems utilizing Western blots

 Transfectants introduced to antibiotic containing HS media produced viable cells containing inserted gene of interest in genome

Figure 13: Transcriptional Control using Various Inducible Promoters (A & B) or Constitutive Promoters (C & D)

A & B

Tetracycline inducible promoter system
TetR binds to *tetO* operon preventing transcription factors binding

Introduction of tetracycline releases tetO, promoting transciption

C & D

C: High expression --> 44,000 Miller Units D: Low expression --> 2,500 Miller Units



Figure 14: Diagram of Isoprene Applications



Figure 15: Translcriptional control via UAG codon

Encodes for stop as well as pyrrolysine

• Subsequent nucleotides determine stop read or pyrrolysine

A or T--Stop codon

read

G or C--Pyrrolysine

• Transcriptional control tested against reporter gene, uidA

Characterization of pyrrolysine control using Western blots

Carbonates

• Single-cell genome sequencing

- Metagenomics
- Meta-transcriptomics

Mechanism of carbonate dissolution

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